

COURSE DESCRIPTION CARD

Faculty of Civil Engineering and Environmental Sciences									
Field of study								Degree level and programme type	
Specialization/ diploma path	-							Study profile	Academic profile
Course name	Sanitary chemistry							Course code	IS-FCEE-00268W
								Course type	Erasmus
Forms and number of hours of tuition	L	C	LC	P	SW	FW	S	Semester	winter
	15		30					No. of ECTS credits	4
Entry requirements	Basic knowledge of chemistry, sanitary biology, water management and water protection, water and wastewater technology								
Course content	To familiarize students with the chemical structure of the water molecule, chemical bonds, and resulting properties. Introduce students with natural water and wastewater chemistry, chemical reactions, parameters affecting their course, disturbing factors and methods of their elimination. To introduce and familiarize students with the cycles of major elements in the environment. Learning the correct classification of natural waters based on water quality indicators (surface and groundwater). The basic definitions and classifications of pollution. To teach the most important water and wastewater laboratory analysis methods, which are the basis for the selection of technological processes. To familiarize students with methods of verification and interpretation of obtained test results and their statistical elaboration.								
Teaching methods	Lectures, Laboratory classes (working independently, in pairs or as part of a small team)								
Assessment method	discussion of obtained research results during classes; written reports in research; written tests checking the learning outcomes								
Symbol of learning outcome	Learning outcomes								Reference to the learning outcomes for the field of study
LO1	A graduate knows phenomena, processes and objects which are the basis for identifying sources of water pollution; has the knowledge of physico-chemical and biochemical transformations taking place in natural waters and sewage; can correctly draw conclusions								IS1_W01 IS1_W02 IS1_W07 IS1_U14
LO2	A graduate knows and understands the phenomena, processes occurring in water and sewage; can analyze, evaluate and predict the consequences of the presence of pollutants and toxic substances in them								IS1_W02 IS1_W07 IS1_U08
LO3	A graduate knows and understands the issues of basic laboratory techniques used in environmental engineering; knows the methodology of physico-chemical analyses which are the basis for the assessment of water and wastewater quality;								IS1_W02 IS1_W07 IS1_U02

	knows how to carry out the tests respecting the principles of safety and hygiene	IS1_K07	
L04	A graduate has knowledge and is able to use theoretical fundamentals of chemistry to write about chemical reactions, calculations necessary in the course of research and their interpretation	IS1_W02 IS1_U01	
L05	A graduate has knowledge and is able to use the literature, legal acts, subject standards, databases in order to draw conclusions, conduct experiments properly, as well as to evaluate the verification and interpretation of research results obtained	IS1_W07 IS1_U14	
L06	A graduate is ready to analyze content from a variety of sources and to critically evaluate his/her knowledge.	IS1_K01	
Symbol of learning outcome	Methods of assessing the learning outcomes	Type of tuition during which the outcome is assessed	
L01	discussions during classes, written tests, exercise reports, oral or written colloquia	L, LC	
L02	discussions during classes, written tests, exercise reports, oral or written colloquia	L, LC	
L03	discussions during classes, written tests, exercise reports, oral or written colloquia	L, LC	
L04	discussions during classes, written tests, exercise reports, oral or written colloquia	L, LC	
L05	discussions during classes, written tests, exercise reports, oral or written colloquia	L, LC	
L06	discussions during classes, written tests, exercise reports, oral or written colloquia	L, LC	
Student workload (in hours)		No. of hours	
Calculation	Lecture attendance	15	
	participation in classes	30	
	preparation for classes, projects, seminars, etc.	10	
	working on projects, reports, etc.	20	
	participation in student-teacher sessions related to the classes/seminar/project	15	
	implementation of project tasks	5	
	preparation for and participation in exams/tests	20	
	TOTAL:	115	
Quantitative indicators		HOURS	No. of ECTS credits
Student workload – activities that require direct teacher participation		50	2
Student workload – practical activities		95	3,8

Basic references	1.Manahan S.E. Environmental Chemistry. Taylor & Francis/CRC Press, 2009; 2.Andrews J.E., Brimblecombe P., Jickells T.D., Liss P.S., Reid B. J. An Introduction to Environmental Chemistry. Blackwell Publishing, 2004 3.Miroslav Radojevic Vladimir N Bashkin; V. N: Practical environmental analyses, Royal Society of Chemistry (Great Britain), 2006; 4.R M Harrison, P Monks; Stephen J De Mora; J. G Farmer; M. C Graham; C Hulsall; Ian D Pulford: Principles of environmental chemistry, Society of Chemistry (Great Britain) 2007	
Supplementary references	Crowe J., Bradshaw T., Chemistry for the Bioscience, Oxford University Press, Oxford, 2010.	
Organisational unit conducting the course	Department of Technology in Environmental Engineering	Date of issuing the programme
Author of the programme	Assoc. Prof. Joanna Szczykowska, PhD	23.03.2023

L – lecture, C – classes, LC – laboratory classes, P – project, SW – specialization workshop, FW - field work,

S – seminar